## NMR SPECTRA **OF CARDENOLIDES WITH AN** OXYGEN-CONTAINING FUNCTION AT C<sub>10</sub>

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In the majority of papers devoted to the NMR-spectroscopic study of the cardenolides [1-8], with a few exceptions [1-2], only steroid aglycones with methyl groups at  $C_{10}$  and  $C_{13}$  are considered. We give the results of an NMR-spectroscopic investigation with some cardenolides with oxygen-containing functions at  $C_{10}$ .

Chemical shifts of the angular  $C_{18}$  methyl group. If literature data on the acetates of periplogenin and strophanthidin  $[4-8]$  and the results that we have obtained for strophanthidin (I), strophanthidol (III), and  $10\beta$ -hydroxyl-19-norperiplogenin (VIII) are compared, it can be seen that the values of the chemical shifts for the 18-methyl protons are almost the same in all cases, amounting to 9.10-9.13  $\tau$ . This means that substituents at C<sub>10</sub> practically do not interact with the 18-methyl protons and do not affect the position of their signal in the spectrum. The introduction of a  $\Delta^5$  bond into the strophanthidin molecule causes slight screening of the C<sub>18</sub> methyl group. Thus, for example, the signal of the  $C_{18}$  methyl group in pachygenin (IV) (table) shifts into the stronger field by 0.06 ppm ( $\Delta \tau = \tau_{IV} - \tau_{I} = 0.06$  ppm). As was to be expected [6], in diffugenin (VI), because of the presence of a  $\Delta^{14}$  bond, the screening effect appears far more ( $\Delta \tau = \tau_{VI} - \tau_I = 0.25$  ppm). The trans linkage of the A and B rings and the absence of a hydroxyl group at C<sub>5</sub> in corotoxigenin (V) leads to only a small displacement of the signal of the C<sub>18</sub> methyl group to the stronger field ( $\Delta \tau = \tau \gamma$  - $- \tau_{I} = 0.08$ ).

The introduction of an OH group into the 17 $\alpha$  position has a considerable influence on the chemical shift of the  $C_{18}$  methyl group. Thus, in the 19-norpentahydroxycardenolide X and its acetate XI and in the acetate of 17 $\alpha$ -hydroxystrophanthidin (XII), the chemical shift of the  $C_{18}$  methyl changes in the narrow range of 8.95-8.88 ppm and is displaced into the weaker field as compared with the field of the  $C_{18}$  methyl of strophanthidin by 0.18-0.24 ppm.

Signals of the protons of the butenolide ring. The protons of the  $C_{21}$  methylene group of the butenolide ring form an AB system which interacts weakly with the vinyl proton at  $C_{22}$  [2]. In the spectra of strophanthidin (I) and its acetate (II) and of strophanthidol (III) and 10 $\beta$ -hydroxy-19-norperiplogenin (VIII) and its acetate (IX), the signals of the C<sub>21</sub> methylene protons appear in the form of a well-resolved quartet. It follows from the figures in the table that a change in the substituent at  $C_{10}$  is not reflected in the chemical shift of the  $C_{21}$  protons. The introduction of a double bond in the  $C_5$  position (IV) or a change in the linkage of the A and B rings (V) also has little effect on the shift of the signals of the  $C_{21}$  protons. However, a double bond located at  $C_{14}$  (VI) and (VII) shifts the signals of the  $C_{21}$  protons in the high-field direction by 0.28 and 0.88 ppm. At the same time, the nature of the signals of the AB quartet of these compounds shows that the chemical shifts of the  $C_{21}$  protons are extremely close.

A 17 $\alpha$ -hydroxyl descreens the C<sub>21</sub> protons, which shifts their signal into a weaker field.

The signal of the  $C_{22}$  vinyl proton, which generally appears in the form of a poorly resolved triplet [6], resonated in the  $4.02-4.06$   $\tau$  region in the cardenolides that we studied. It follows from the table that only the introduction of a  $17\alpha$ -hydroxyl (compounds X-XIII), which displaces this signal to the weaker field by 0.56-0.58 ppm, has a marked effect on the shift of this proton.



Signal of the proton at C<sub>17</sub>. It is known [4, 6] that the signal of the proton at C<sub>17</sub> appears in the form of a broad multiplet and its chemical shift for the 19-methylcardenolides is  $7.14-7.17$   $\tau$ . In the cardenolides that we studied, the  $17\alpha$ -proton resonates at 7.32-7.37  $\tau$ . The position of its signal is almost independent of the nature of this substituent at  $C_{10}$ , the linkage of the A/B rings, and the presence of double bonds at  $C_5$  and  $C_{14}$ .

Signals of the protons of a  $C_{19}$  oxygen-containing substituent. As was to be expected, in all the compounds with a carbonyl function at  $C_{10}$  (I, II, IV-VII, XII), the aldehyde proton resonates in a very weak field in the form of a sharp



NMR Spectra of Cardenolides with Oxygen-Containing Functions at C<sub>10</sub> (7 scale).

\*Broadened singlet<br>\*\*Rings A and D in the trans-linkage<br>s) singlet; d) doublet; m) multiplet; q) quadruplet.

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singlet. The formation of a  $\Delta^{14}$  bond or even the introduction of a 17 $\alpha$ -hydroxy group has no appreciable influence on the position of this signal. Conversely, a double bond at  $C_5$  and a change in the linkage of the A/B ring (compounds IV and V) shift the signal into a weaker field by  $0.57$  and  $0.28$  ppm, respectively.

It is an interesting fact that in the NMR spectrum of strophanthidol (III), because of the nonequivalence of the methylene protons of the C<sub>19</sub>-CH<sub>2</sub>OH group, a one-proton quadruplet is found at 6.23  $\tau$ (J<sub>1</sub> = 10 Hz and J<sub>2</sub> = 3 Hz) and also a one-proton doublet at 5.36  $\tau$  (J = 10 Hz). When the signals of the hydroxy groups are displaced by means of  $CF<sub>3</sub>COOH$ , the methylene protons of the  $C<sub>19</sub> - CH<sub>2</sub>OH$  group form a well-defined AB system.

Signals of the protons at  $C_3$ . In the NMR spectra of the acetates of the compounds that we studied, the protons of the acetyl groups form a narrow singlet at 8.11-8.17  $\tau$ . In the cardenolides themselves, the proton at C<sub>3</sub> attached to the same carbon atom as the hydroxyl group is generally descreened and is located in the 5.7-6.3  $\tau$  region. On passing to the acetates, it undergoes a paramagnetic shift by 0.9-1.0 ppm [2]. It is known [2] that axial protons attached to the same atom of carbon as an acetate or hydroxyl group resonate in a stronger field than the corresponding equatorial protons in the epimeric compounds. As can be seen from the table, in pachygenin (IV) and corotoxigenin (V) the  $3\alpha$ -axial proton resonates in a stronger field than the  $3\alpha$ -equatorial protons in the  $5\beta$ -cardenolides. Moreover, the  $3\alpha$ -axial protons in compounds IV and V are subject to diaxial and axial-equatorial interaction with the protons at  $C_2$  and  $C_4$ , which leads to a very broad signal with  $\Delta f_{1/2}$  = 15-23 Hz. Conversely, the signal of the 3 $\alpha$ -equatorial proton is fairly narrow and sharp  $(\Delta f_{1/2} = 8-9 \text{ Hz})$ .

Signals of the protons at the  $\Delta_5$  and  $\Delta_{14}$  bonds. In pachygenin (IV), the resonance signals of the proton at C<sub>6</sub> appears at  $4.3\pi$  and has the form of a broad multiplet. The olefinic proton at C<sub>15</sub> in compounds VI and VII resonates in the 4.84--4.87 7 region in the form of a signal with a half width of 7 Hz, which shows its weak interaction with the protons at  $C_8$  and  $C_{16}$ .

## Experimental

The spectra were recorded on a JNM-4H-100 instrument with a working frequency of 100 MHz in deuteropyridine. HMDS was used as the internal standard, its signal being taken as 10 ( $\tau$ -scale). The displacement of the signals of the hydroxyl group was made with the aid of  $CF_3COOH$ .

## Conclusions

The NMR spectra of 12 cardenolides with oxygen-containing functions at  $C_{10}$  has been studied. The chemical shifts of the signals of the protons of the C<sub>18</sub> angular methyl group, a butenolide ring, and some other protons have been discussed as functions of the structural features of the individual compounds.

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